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EXCHANGE OF METAL CARBONYL MOIETIES ON GERMANES

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Summary

For $R_3Ge = GeH_3$, CH_3GeH_2 or $(CH_3)_2GeH$, $R_3GeCo(CO)_4$ reacts in an ether solvent with $Mn(CO)_5$ to give $R_3GeMn(CO)_5$ while $Mn(CO)_5$ displaces $Fe(CO)_4^2$ or $(GeH_3)Fe(CO)_4$ and $Fe(CO)_4^2$ displaces $Co(CO)_4$.

Group IV derivatives of metal carbonyls are commonly made [1] by the halide elimination reaction, see eq. 1 for example [2]. We now report an

$$GeH_3Br + Co(CO)_4 \rightarrow GeH_3Co(CO)_4 + Br$$

extension of this reaction, where one metal carbonyl anion displaces another on germanium (eq. 2). Such transition metal exchange processes are rare: one

$$CH_{3}GeH_{2}Co(CO)_{4} + Mn(CO)_{5} \rightarrow CH_{3}GeH_{2}Mn(CO)_{5} + Co(CO)_{4}$$
⁽²⁾

example [3] involved the displacement of $(C_5H_5)Mo(CO)_3^{-1}$ from carbon by $(C_5H_5)Fe(CO)_2^{-1}$.

Reactions such as 2 also occur with GeH₃ and $(CH_3)_2$ GeH derivatives. At room temperature they take 5 to 15 minutes in ether solvents on a 0.2 to 0.5 mmol scale using sodium salts of the carbonyl anion. No reaction occurs in hydrocarbons or in absence of a solvent. The products were characterised by comparison with authentic samples [4].

Using equimolar quantities, the yields of germylmanganese compounds were about 80% after purification, suggesting (2) is near-quantitative. This was confirmed by adding HCl to the residues which yielded $HCo(CO)_4$ with only traces of $HMn(CO)_5$.

A similar reaction between $Mn(CO)_5$ and $(GeH_3)_2Fe(CO)_4$ in Et_2O or THF gave $GeH_3Mn(CO)_5$ corresponding to 43% of the GeH_3 groups, together with traces of $(GeH_3)HFe(CO)_4$. Addition of HCl to the residues yielded some $HMn(CO)_5$ together with $(GeH_3)HFe(CO)_4$ but little $H_2Fe(CO)_4$. The principal reaction is therefore:

(1)

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$Mn(CO)_{5}$ + $(GeH_{3})_{2}Fe(CO)_{4} \rightarrow GeH_{3}Mn(CO)_{5}$ + $GeH_{3}Fe(CO)_{4}$

Preliminary studies indicate $GeH_3Co(CO)_4$ exchanges with $Fe(CO)_4^{2^-}$ to yield $(GeH_3)_2Fe(CO)_4$.

Taken together, these results indicate a metal carbonyl displacement series on GeH₃ and related germyls in the sense $Co \rightarrow Fe \rightarrow Mn$. If this series can be extended to other metal carbonyls and related species it will provide a useful new synthetic method. Taken in combination with reported Group IV displacements on a given transition metal such as [5] Me₃Si \rightarrow Me₃Ge \rightarrow Me₃Sn in Me₃MCo(CO)₄, a wide range of interconversions in the Group IV transition metal family is suggested. Of particular interest is the possibility of mixed ligand iron compounds (R₃M)(R'₃M')Fe(CO)₄ and as a route to new systems: for example, GeH₃Co(CO)₄ is completely consumed when added to Cr(CO)₅^{2⁻}, though Ge—Cr species have not yet been isolated.

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